

BSc Pharmacology and Translational Medicine

Module: Principles of Pharmacodynamics and Pharmacokinetics
Lecture: Transport of Drugs Across Biological Membranes
Date: Tuesday 23rd October 2012



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Transport of Drugs Across Biological Membranes

Lecture Outline

- Recap on Drugs and Drug Therapy
- Aims of Drug Therapy
- Routes of Administration
- ADMET
- Recap on Structure of Biological Membranes
- Crossing Cell Membranes
- Diffusion
- Carrier-Mediated Transport
 - Facilitated Diffusion
 - Active Transport

Drugs and Medicines

- A *drug* can be defined as:

“A chemical substance of known structure, other than a nutrient or an essential dietary ingredient, which, when administered to a living organism, produces a biological effect.”

- Drugs may be synthetic chemicals, chemicals obtained from plants or animals, or products of genetic engineering.
- A *medicine* is a chemical preparation, which usually but not necessarily contains one or more drugs, administered with the intention of producing a therapeutic effect.
- Medicines usually contain other substances (excipients, stabilisers, solvents, etc.) besides the active drug, to make them more convenient to use.
- To count as a drug, the substance must be administered as such, rather than released by physiological mechanisms.

Aim of Drug Therapy

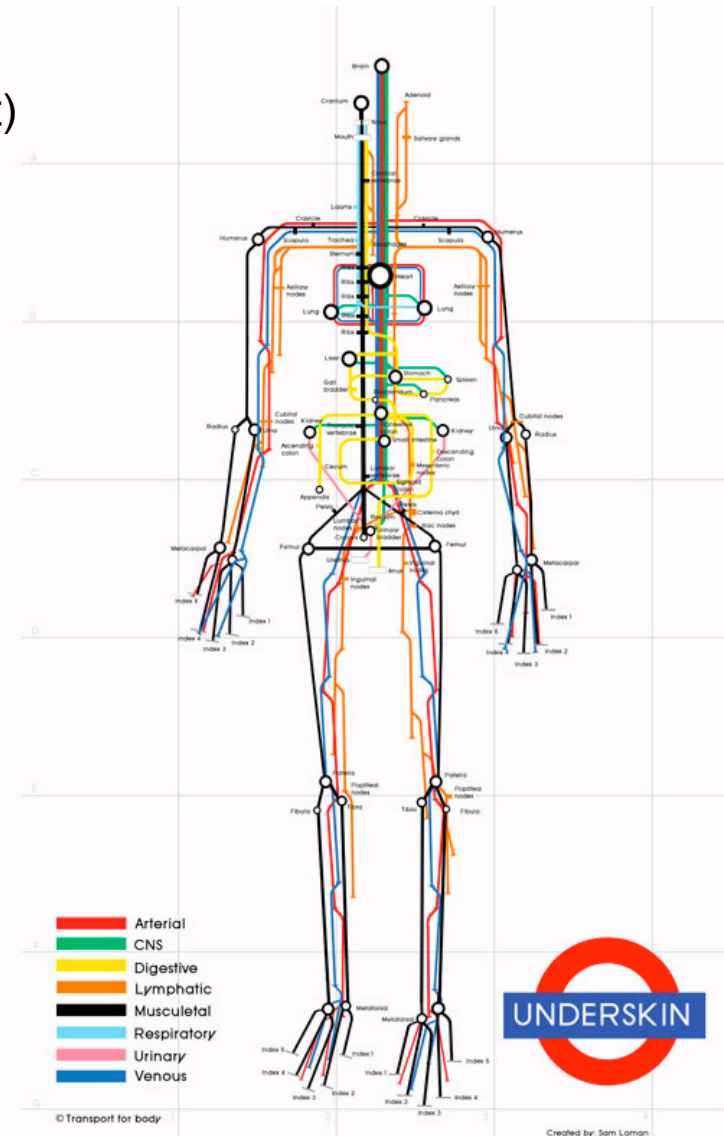
Drug Therapy

- The main aim of drug therapy is to use (bio)chemical agents to treat, prevent or control disease
- Understanding how drugs interact with the body is key to their use in a rational way and for discovering and developing new treatments
- The ideal is to deliver a pharmacologically effective dose to a target without causing (excessive) toxicity or unwanted off-target effects
- A number of factors must be considered in relation to the administration of drugs in therapy:
 - Absorption
 - Distribution
 - Metabolism
 - Elimination
 - (Toxicity)
- ADME(T)

Routes of Administration

Routes of Administration

- **Enteral** (use of gastrointestinal (GI) tract)
 - Oral
 - Sublingual
 - Rectal
- **Parenteral** (not through the GI tract)
 - **Intravascular**
 - Intravenous (IV)
 - Intra-arterial (IA)
 - Intramuscular (IM)
 - Subcutaneous (SC)
- **Other Routes**
 - Inhalation
 - Intranasal
 - Intrathecal / Intraventricular
 - Topical
 - Transdermal



Routes of Administration

Absorption and Distribution

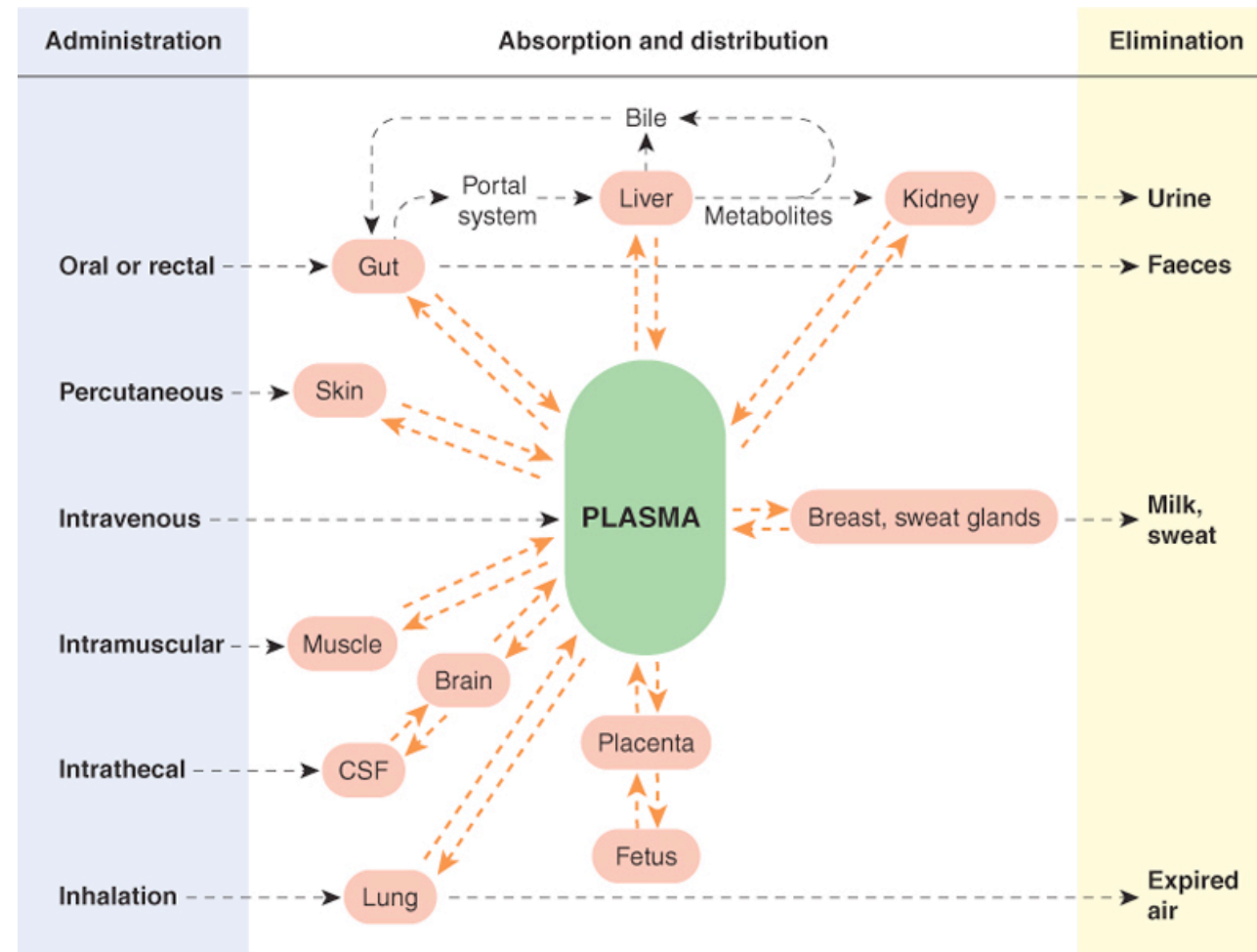
- Blood plasma is the main bulk carrier of drugs around the body and is therefore absorption into the plasma is the primary mechanism for subsequent distribution
- Absorption relates to the movement of a drug from the site of administration into the blood plasma (in most cases)
- The route of administration greatly influences the rate / profile of absorption
 - E.g. oral vs intravenous
- Upon administration, a drug will need to travel to the target in order to bring about the desired therapeutic effect
- Transport between and within the different tissues and compartments is necessary
- Different tissues and compartments have different properties that influence drug transport and partitioning

ADMET

▪ A number of factors must be considered in relation to the administration of drugs in therapy:

- Absorption
- Distribution
- Metabolism
- Elimination
- Toxicity

- Rates
- Routes

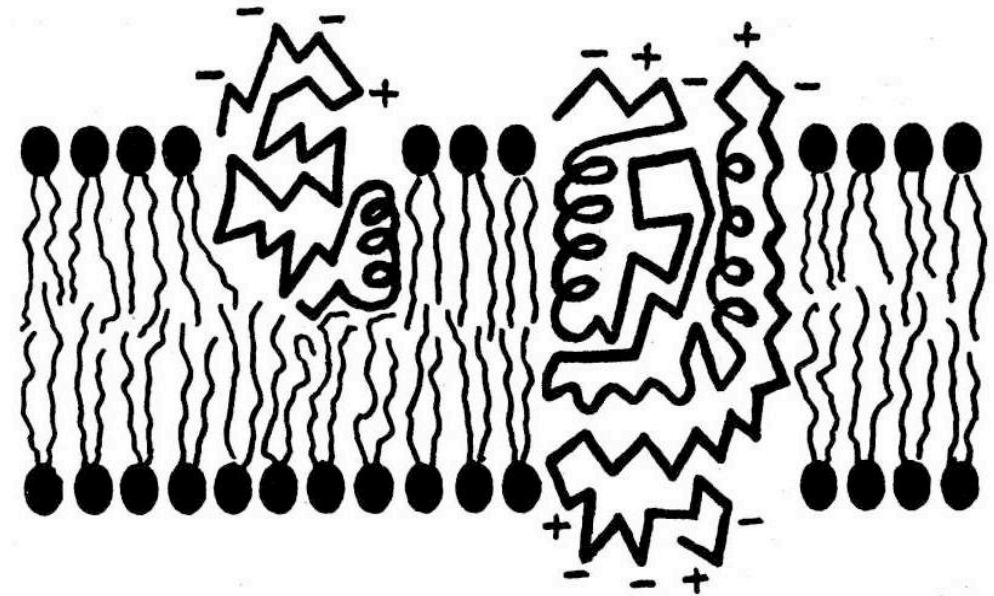
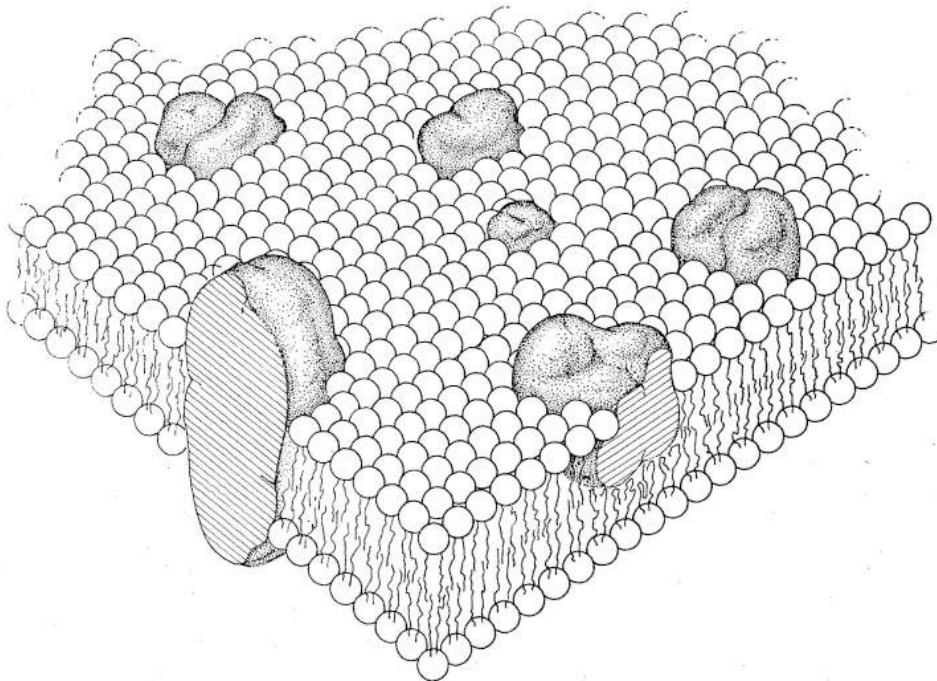


Structure of Biological Membranes

Structure of Biological Membranes

Fluid Mosaic Model

- Singer and Nicolson (1972)
- Fluid Mosaic model
- Currently accepted basic model for cellular membranes

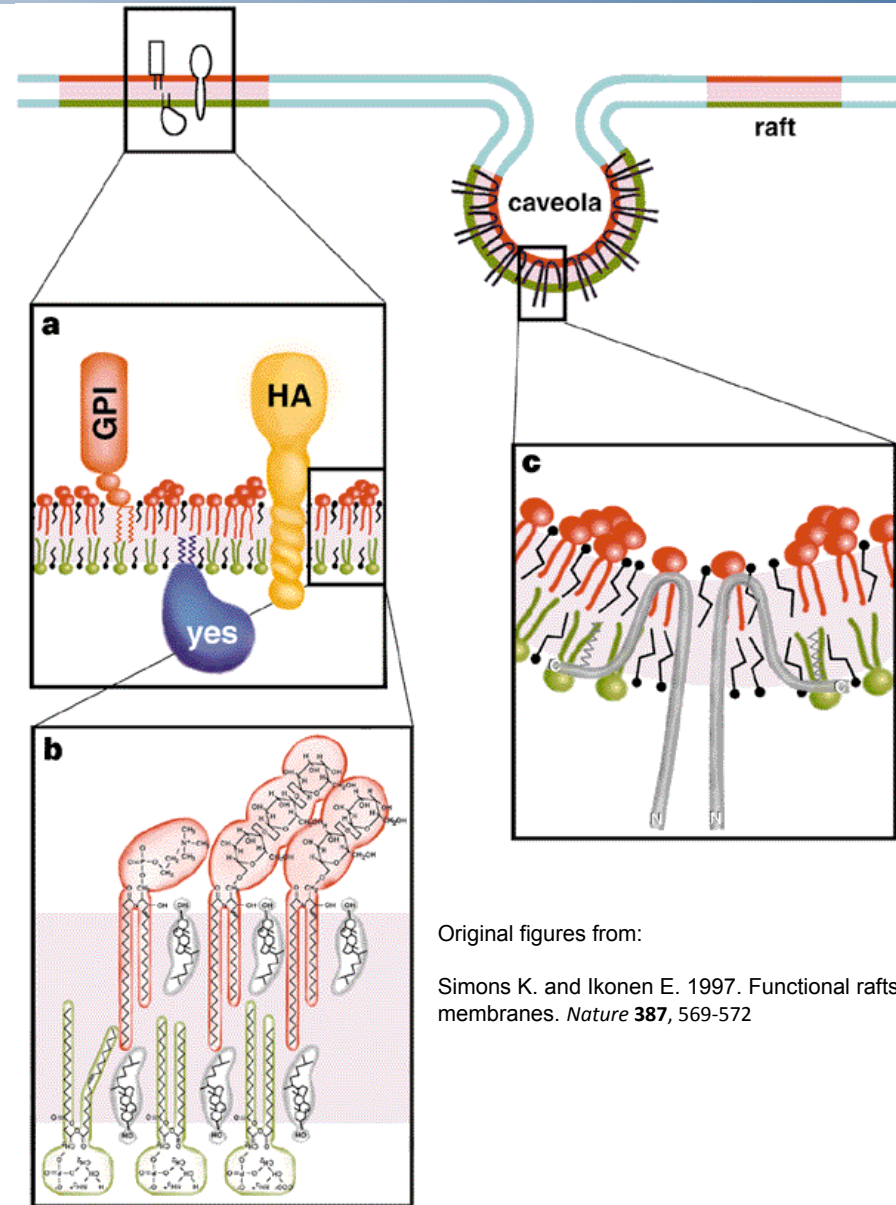


Original figures from: Singer, S.J. and G. L. Nicolson. 1972. The fluid mosaic model of the structure of cell membranes. *Science*. 175: 720-731

Structure of Biological Membranes

Lipid Raft Model

- Simons *et al.* (1997)
- Lipid raft model
- Lipids are asymmetrically distributed between the cytoplasmic and exoplasmic leaflets of the membrane. E.g.:
 - Exoplasmic leaflet enriched in sphingomyelin and glycosphingolipids
 - Cytoplasmic leaflet enriched in phosphatidylserine and phosphatidylethanolamine
- Lipids also exhibit a degree of lateral organisation in microdomains termed lipid rafts
- Proteins may be selectively recruited to rafts depending on their composition
- The function of lipid rafts in the cell membrane is currently a growing area of research



Original figures from:

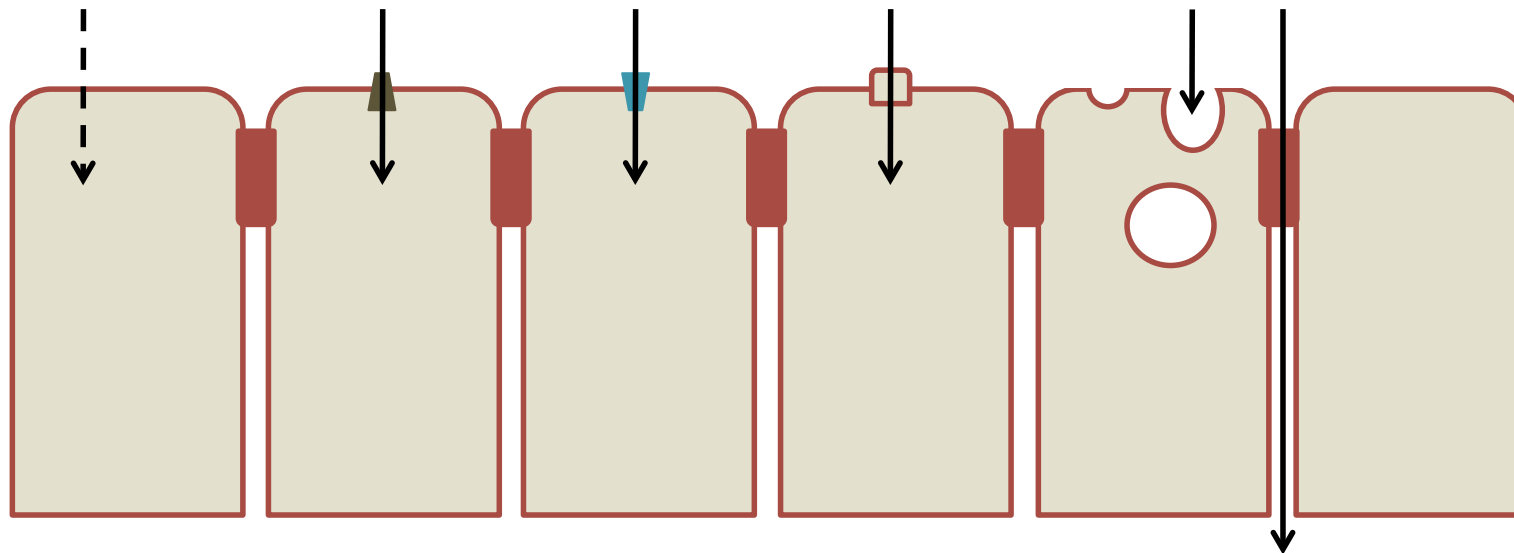
Simons K. and Ikonen E. 1997. Functional rafts in cell membranes. *Nature* **387**, 569-572

Crossing Cell Membranes

Crossing Cell Membranes

Types of Cell Membrane Transport

- There are six main ways for small molecules to cross cell membranes:
 - Passive diffusion through the phospholipid bilayer
 - Facilitated diffusion via transmembrane transport proteins
 - Active transport via transmembrane transport proteins
 - Filtration through aqueous pores in the phospholipid bilayer
 - Pinocytosis
 - Paracellular transport



Crossing Cell Membranes

Passive Diffusion

Diffusion Across Cellular Membranes

Passive Diffusion

- Non-polar substances can be readily dissolved in non-polar solvents
- Therefore they are able to freely diffuse across (and within) the phospholipid cell membrane
- The rate of diffusion across a membrane can be determined by
 - The permeability of the substance in the membrane
 - The concentration gradient across the membrane
- Rapid diffusion across the membrane requires permeant species to be present in the membrane in sufficient numbers and to be freely mobile

Diffusion Across Cellular Membranes

Membrane Permeability

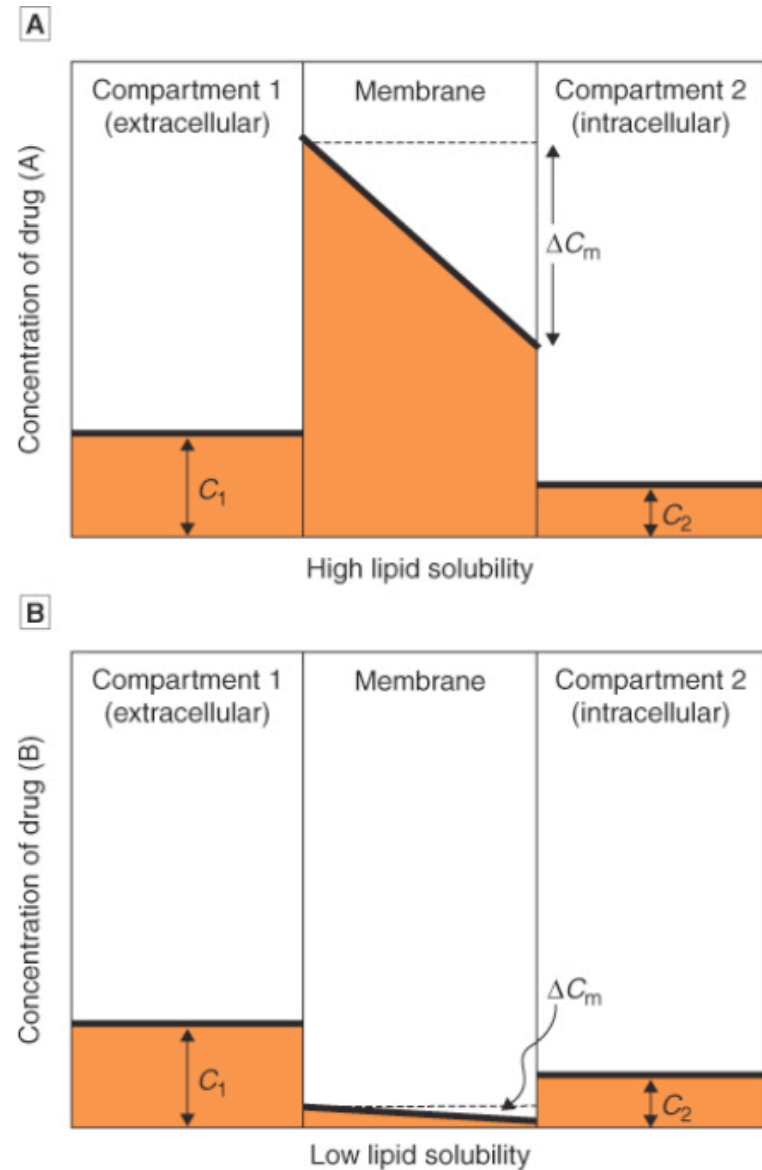
- The permeability of a diffusing species in a cell membrane is largely related to its physicochemical properties, primarily:
 - The diffusivity of the species in the membrane, denoted by the diffusion coefficient.
 - The solubility of the species in the membrane, denoted by the partition coefficient.
- The diffusivity of different small molecules does not vary considerably, and therefore contributes little to the differences observed in membrane permeability between drugs
- Therefore the membrane (lipid) solubility of a molecule is the key factor in determining membrane permeability and rate of diffusion. This physicochemical property is therefore of considerable importance in understanding the transport of drugs
- The usual surrogate for estimating this is the logP octanol-water coefficient, which can be measured or predicted using computational chemistry methods

$$\log P_{oct/wat} = \log \left(\frac{[\text{drug}]_{oct}}{[\text{drug}]_{wat}} \right)$$

Diffusion Across Cellular Membranes

Membrane Permeability

- The transmembrane concentration gradient can be denoted ΔC_m
- ΔC_m increases with increasing partition coefficient (P)
- Therefore, the rate of diffusion increases with increasing P



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Diffusion Across Cellular Membranes

Membrane Permeability – Molecular Weight

- The partition coefficient (P) is by far the greatest indicator of the rate of diffusion through a cell membranes. It contributes greatly to the overall permeability of a substance in a membrane.
- The diffusivity of different small molecules does not vary considerably, and therefore contributes little to the differences observed in membrane permeability between drugs
- However, the rate of diffusion of species with the same P value can vary to some degree
- Molecular weight is a good indicator of the relative diffusivity of a species, and provides an additional property with which to model diffusion in cell membranes
- Molecules up to a Mw of ~200 can diffuse into and with the membrane, with the rate of diffusion anticorrelated with Mw
- Larger molecules (e.g, sucrose; Mw 300) are excluded from the membrane and do not diffuse

Diffusion Across Cellular Membranes

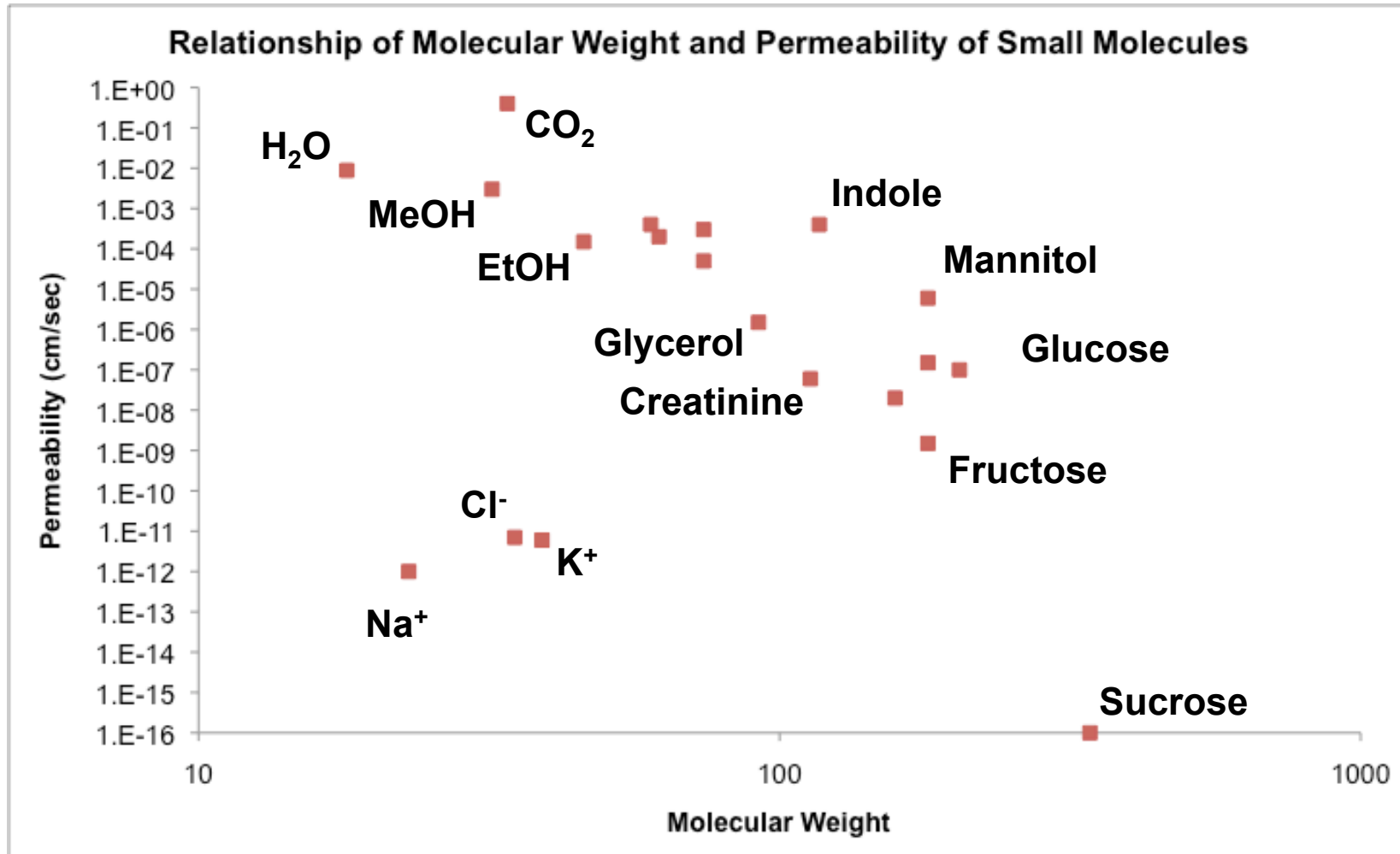
Lipids in Cell Membranes

| Solute | Mw | Permeability (cm/s) |
|-----------------|-----|---------------------|
| Carbon Dioxide | 34 | 4.E-01 |
| Water | 18 | 9.E-03 |
| Methanol | 32 | 3.E-03 |
| Urea | 60 | 4.E-04 |
| Butanol | 74 | 3.E-04 |
| Ethylene glycol | 62 | 2.E-04 |
| Ethanol | 46 | 2.E-04 |
| Methylurea | 74 | 5.E-05 |
| Glycerol | 92 | 2.E-06 |
| Creatinine | 113 | 6.E-08 |
| Indole | 117 | 4.E-04 |
| Uric Acid | 158 | 2.E-08 |
| Glucose | 180 | 2.E-07 |
| Fructose | 180 | 2.E-09 |
| Mannitol | 180 | 6.E-06 |
| Tryptophan | 204 | 1.E-07 |
| Sucrose | 342 | 1.E-16 |
| Na+ | 23 | 1.E-12 |
| K+ | 39 | 6.E-12 |
| Cl- | 35 | 7.E-12 |

Adapted from Saltzman, W.M. Drug Delivery – Engineering Principles for Drug Therapy, OUP, 2001

Diffusion Across Cellular Membranes

Molecular Weight and Ionisation Affect Permeability



Membrane Components

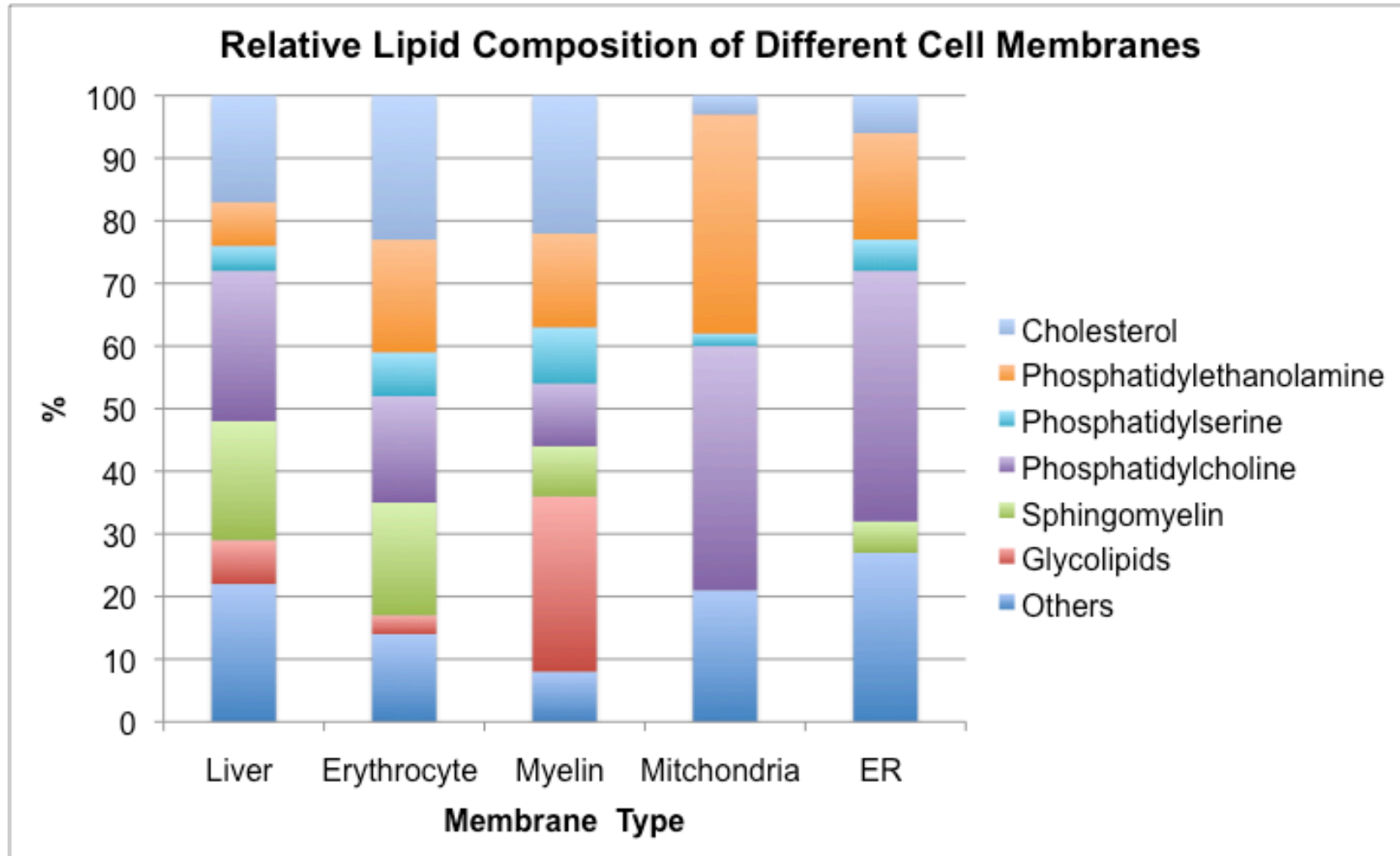
Lipids in Cell Membranes

| | Liver | Erythrocyte | Myelin | Mitochondria | ER |
|--------------------------|-------|-------------|--------|--------------|----|
| Cholesterol | 17 | 23 | 22 | 3 | 6 |
| Phosphatidylethanolamine | 7 | 18 | 15 | 35 | 17 |
| Phosphatidylserine | 4 | 7 | 9 | 2 | 5 |
| Phosphatidylcholine | 24 | 17 | 10 | 39 | 40 |
| Sphingomyelin | 19 | 18 | 8 | 0 | 5 |
| Glycolipids | 7 | 3 | 28 | 0 | 0 |
| Others | 22 | 14 | 8 | 21 | 27 |

Adapted from Frye, L. and Edidin. The rapid intermixing of cell surface antigens after formation of mouse-human heterokaryons. *Journal of Cell Science*, 1970, 7, 319-335

Membrane Components

Lipids in Cell Membranes

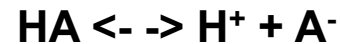


Data from Frye, L. and Edidin. The rapid intermixing of cell surface antigens after formation of mouse-human heterokaryons. *Journal of Cell Science*, 1970, 7, 319-335

pH and Ionisation

Acids and Bases

- The Arrhenius theory of acid dissociation (an acid is a substance that dissociates, releasing H^+)



- Brønsted-Lowry acid-base theory (generalised to proton exchange)



- Acid dissociation constant (K_a) describes the equilibrium position of the dissociation and therefore the strength of the acid.



Svante Arrhenius
(1859 – 1927)



Thomas Martin Lowry
(1874 - 1936)



Johannes Nicolaus Brønsted
(1879 - 1947)

Henderson-Hasselbalch Equation:

$$K_a = \frac{[H^+][A^-]}{[HA]}$$

$$\log_{10} K_a = \log_{10} \left(\frac{[H^+][A^-]}{[HA]} \right)$$

$$\log_{10} K_a = \log_{10} [H^+] + \log_{10} \left(\frac{[A^-]}{[HA]} \right)$$

if $[A^-] = [HA]$

$$-pK_a = -pH + \log_{10} \left(\frac{[A^-]}{[HA]} \right)$$

$$pH = pK_a$$

$$pH = pK_a + \log_{10} \left(\frac{[A^-]}{[HA]} \right)$$

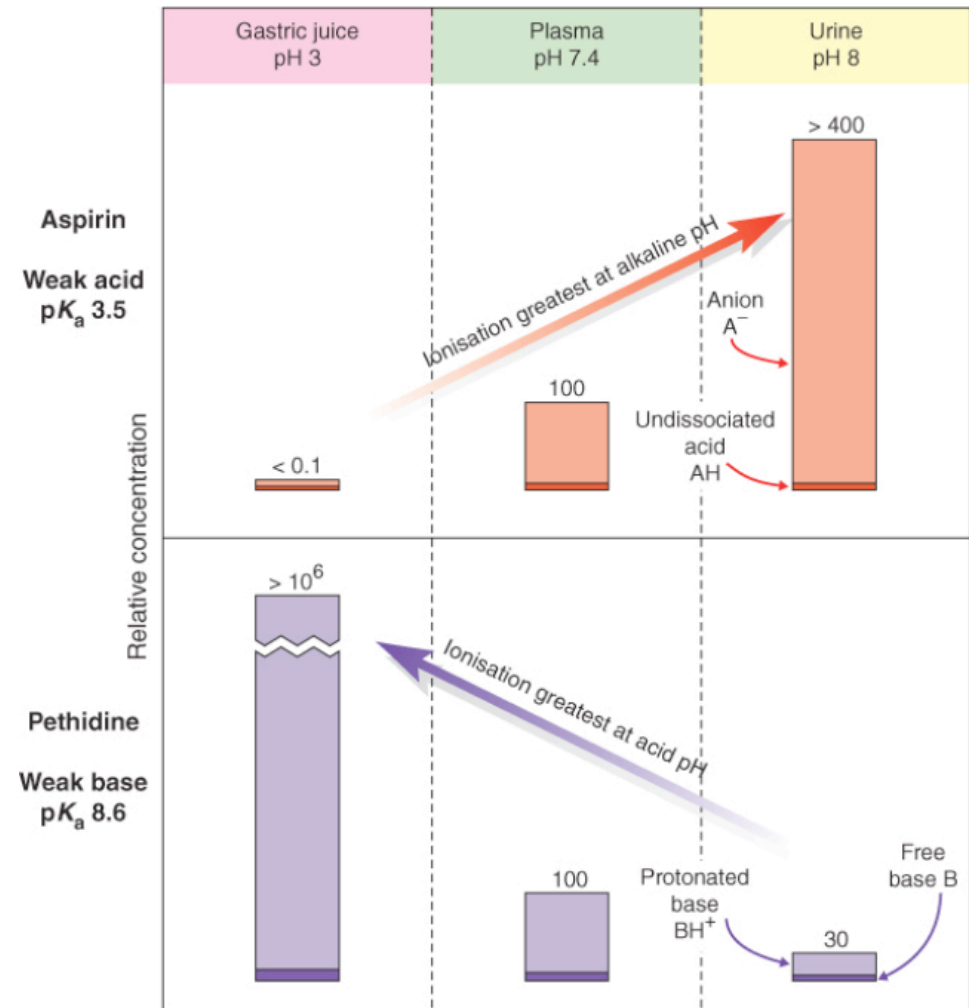
pH Affects Membrane Permeation

- The effect of pH and ionisation on diffusion through cell membranes is often considerable for drugs that are weak acids or bases
- The ionised form will have a very low solubility in lipid, and therefore exhibit very low diffusion/permeation of the phospholipid bilayer
- In some cases the uncharged species is still insufficiently lipid soluble to diffuse
 - e.g. aminoglycosides
- This is usually because of the presence of groups that form hydrogen bonds
 - e.g. $-OH$

pH and Ionisation

pH Partitioning

- pH partitioning may result from 'ion trapping'
- The differing pH of different biological tissues and fluids can result in partitioning where there are differences in the drug concentration
- The steady-state equilibrium position of charged and uncharged species will differ with pH for drugs that are weak acids and bases
- The ratio of the ionised and unionised species in each partition will differ as a result of the pH
- Therefore the total (ionised + unionised) drug concentration will differ
- The ideal (modelled) behavior of partitioning partially helps understand observed behavior



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pH and Ionisation

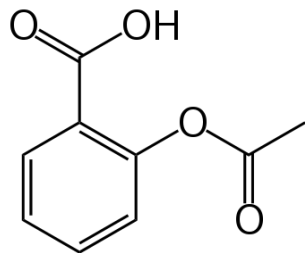
pH Partitioning (2)

Example:

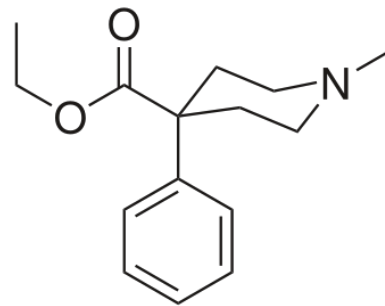
- aspirin (weak acid, $pK_a = 3.5$)
- pethidine (weak base, $pK_a = 8.6$)

Diffusion will result in a constant ratio of the unionised (diffusing) species *between* the partitions

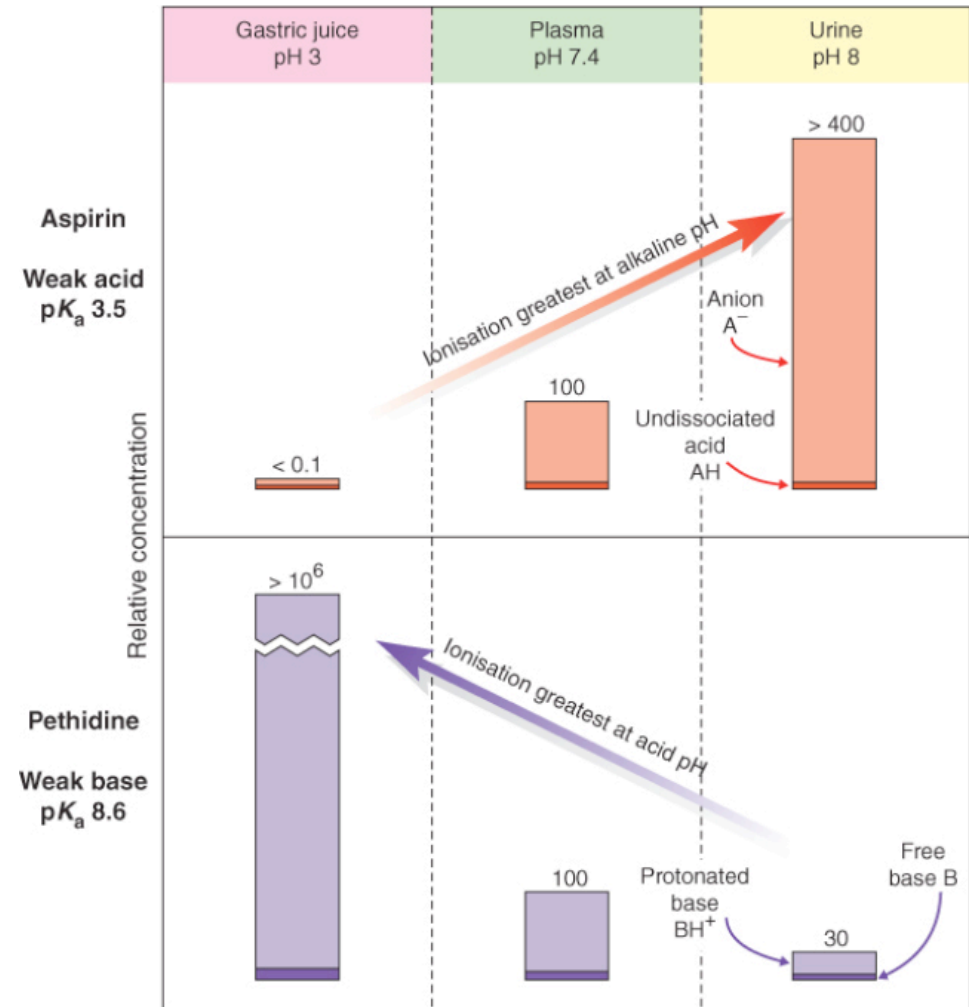
The differing pK_a of the two drugs results in a different distribution of *total* concentration



Aspirin



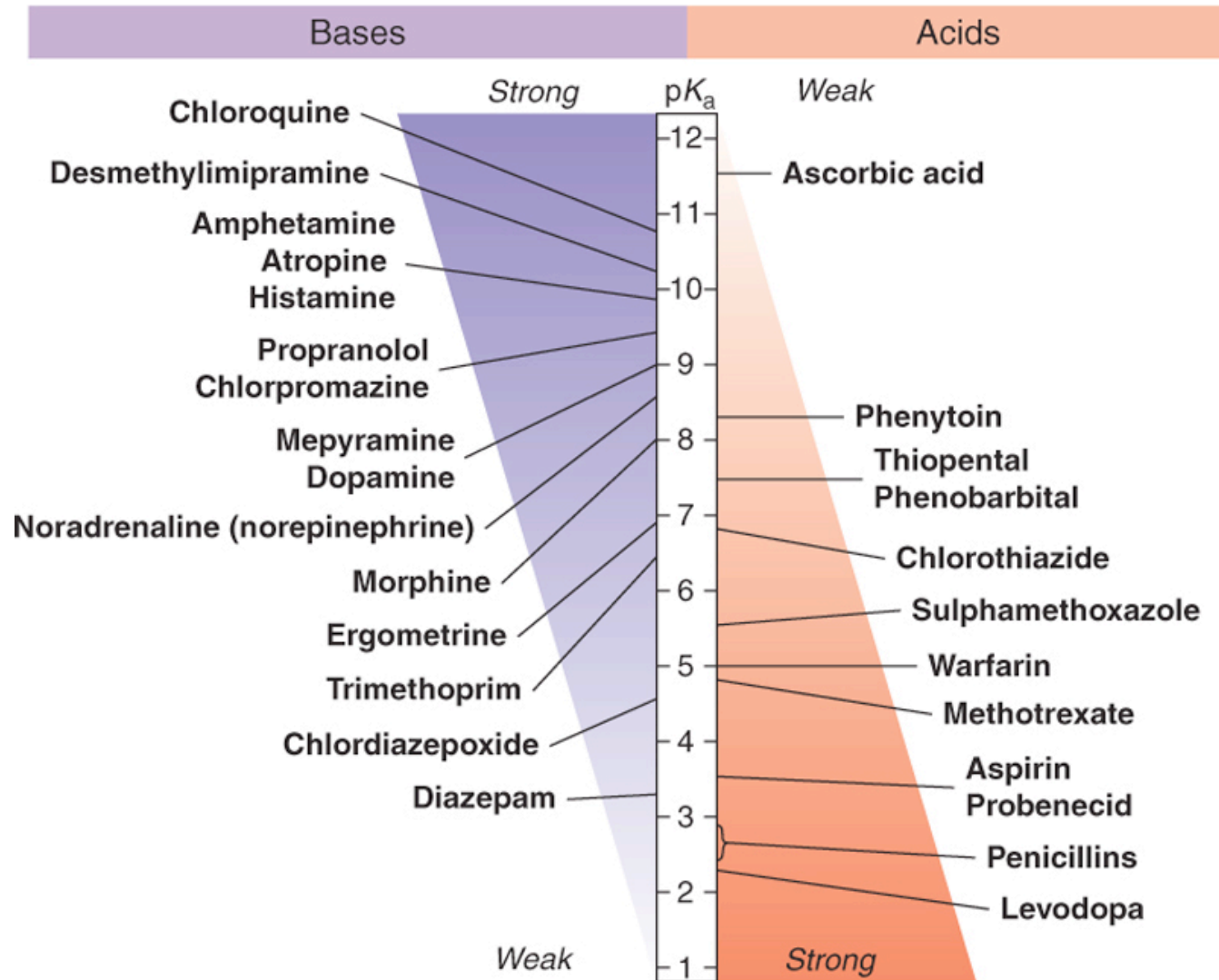
Pethidine



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pH and Ionisation

pKa Values of Common Drugs



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Carrier-Mediated Transport

Carrier-Mediated Transport

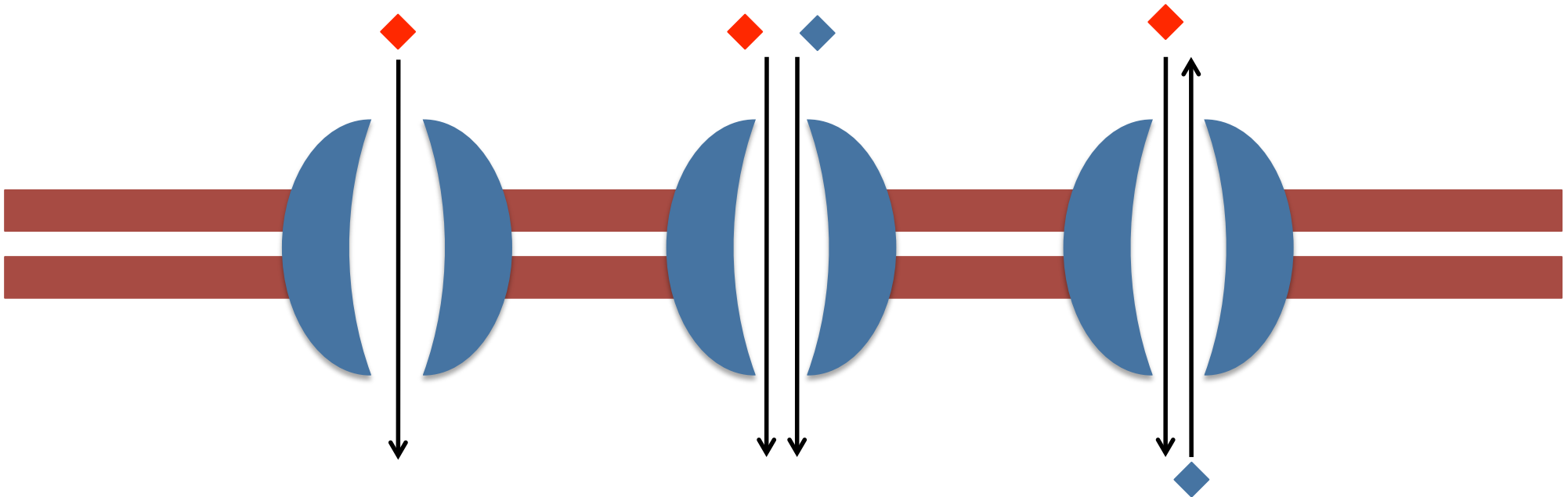
Introduction

- In addition to diffusion, transport across membranes can occur by carrier-mediated processes:
 - Facilitated diffusion
 - Active transport
- Proteins in the membrane facilitate the transit of ions and molecules.
 - The range of molecules that undergo transport depends on the type of transporter
 - Some transporters are more substrate-specific than others

Carrier-Mediated Transport

Transport Processes

- It is useful to describe transporters according to the net translocation of substrates
 - Uniport – the movement of a single molecule at a time
 - Symport – the movement of two molecules in the same direction simultaneously
 - Antiport – the movement of two molecules in opposite directions simultaneously



Carrier-Mediated Transport

Facilitated Diffusion

Carrier-Mediated Transport

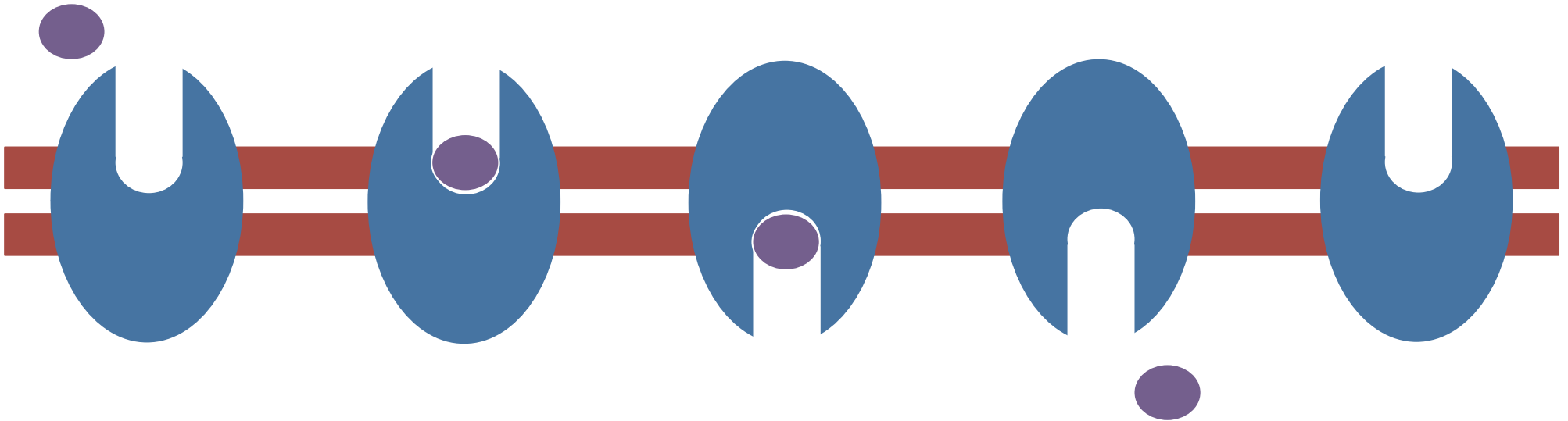
Facilitated Diffusion

- The model of non-mediated transport by diffusion through cellular membranes does not appear to hold for some molecules and ions
- The permeability of some molecules (and ions) is much higher than would be expected
- Transport proteins in the membrane allow transport of molecules *via* different mechanisms and provide an alternative/additional route to diffusion through the membrane
- There are three main types of transport protein that facilitate passive diffusion:
 - Gated pores
 - Ligand-gated ion channels
 - Voltage-gated ion channels

Carrier-Mediated Transport

Facilitated Diffusion – Gated Pores

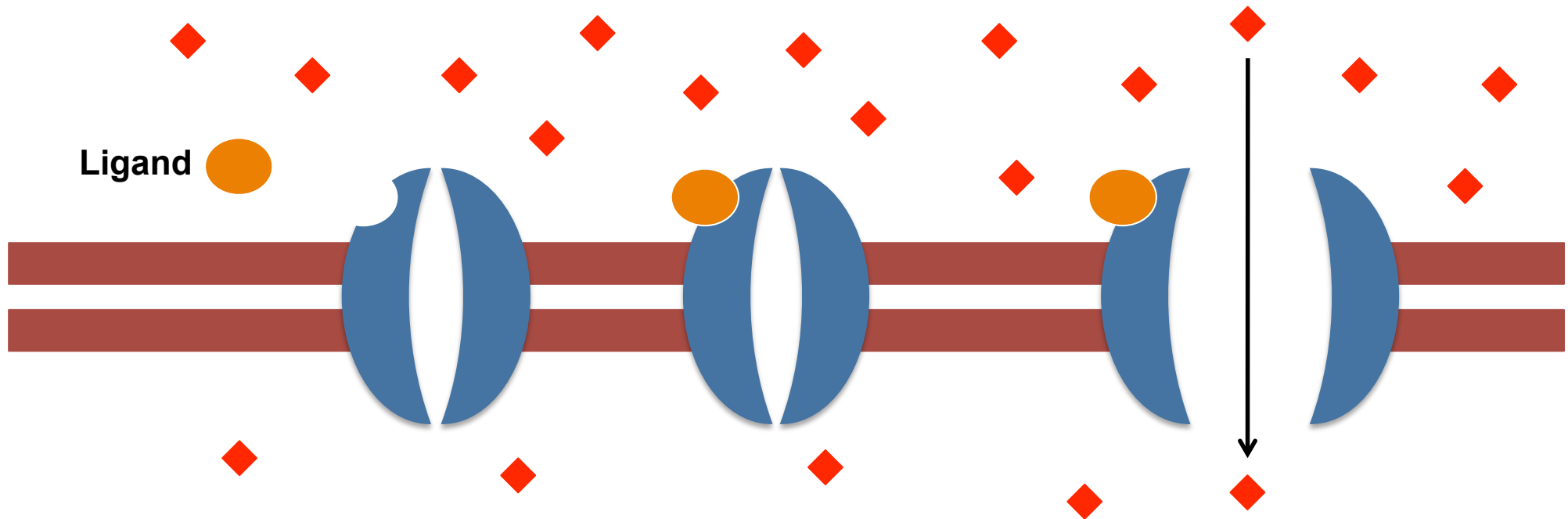
- Example: Glucose transporter
 - Part of the solute carrier (SLC) superfamily of membrane transport proteins
- Glucose binds to the transporter
- The binding of glucose causes a conformational change in the transporter
 - Closes the site of entry
 - Opens the site of exit
- Glucose dissociates from the transporter
- The transporter undergoes conformational change (recovery) to return to the original state



Carrier-Mediated Transport

Facilitated Diffusion – Ligand-Gated Ion Channel

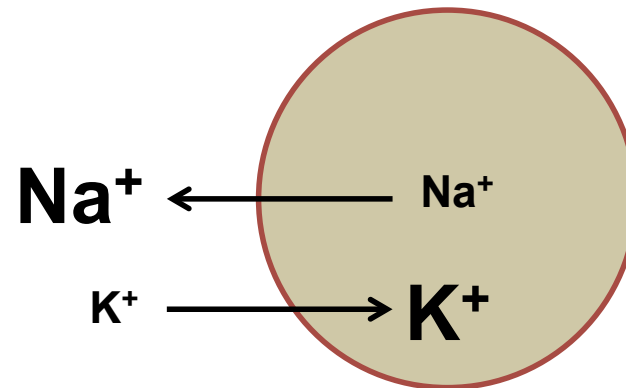
- Some transmembrane proteins alter their conformation when bound with a ligand (usually a small molecule)
- On binding with an appropriate ligand, the protein opens up a channel that is pervious to aqueous species (e.g. ions) that would not normally diffuse through the membrane.



Carrier-Mediated Transport

Facilitated Diffusion – Voltage-Gated Ion Channel

- Some ion channels change confirmation depending on the potential across the plasma membrane
- Example: Neurons and nerve impulses
- In a resting neuron, the interior of the axon is maintained with:
 - a net negative charge
 - a lower concentration of sodium ions
 - a higher concentration of potassium ions



Carrier-Mediated Transport

Facilitated Diffusion – Voltage-Gated Ion Channel

- Sufficient depolarisation and opening of voltage-gated sodium ion channels in the membrane at a point causes adjacent voltage-gated sodium ion channels to open
- On opening, the channels allow influx of sodium ions down the electrochemical gradient to the interior
- This causes further local depolarisation of the membrane and the propagation of the impulse
- The net result of this is a ‘wave’ of depolarisation that travels along the cell (a nerve impulse)
- The mechanism results in a non-attenuated signal transmission along the neuron
- The resting state is restored (after ~1 ms) by proteins known as sodium/potassium ATPase
 - Pump two K^+ into the cell Pump three Na^+ out of the cell
 - Net result is restored polarisation of the membrane

Active Transport

Carrier-Mediated Transport

Active Transport

- Active transport of species across membranes requires the usage of high-energy
- Most typically the energy is provided by the hydrolysis of ATP
- *Indirect active transport* – the mechanism uses the chemical gradient that has already been acquired (by active transport of ions). The energy required to move a chemical species against its own chemical gradient is harnessed by the facilitated diffusion of the ions down their own gradient
- *Direct active transport* – transport proteins use ATP to move chemical species against the chemical gradient

Carrier-Mediated Transport

Indirect ATP-Driven Active Transport

- Indirect active transport occurs by symport of antiport proteins. Key examples are:
 - Symport
 - Na^+ / glucose symport transporter (pump)
 - 1 Na^+ ion and 1 glucose molecule move across the membrane
 - Antiport
 - Na^+ / Ca^{2+} exchanger (NCX)
 - 3 Na^+ ions in and 1 Ca^{2+} out

Carrier-Mediated Transport

Direct ATP-Driven Active Transport

- There main ATP-driven transporters are:
 - Na^+ / K^+ ATPase
 - Uses 1 ATP to transport 3 Na^+ ions out and 2 K^+ ions in
 - Ca^{2+} ATPase
 - Uses 1 ATP to transport 1 Ca^{2+} ion out of the cell
 - H^+ / K^+ ATPase
 - Uses 1 ATP to transport 1 H^+ ion out and 1 K^+ ion in
 - ATP Binding Cassette transporters
 - Ligand transport (usually highly specific for each isoform)
 - Uses 1 ATP to transport a single ligand

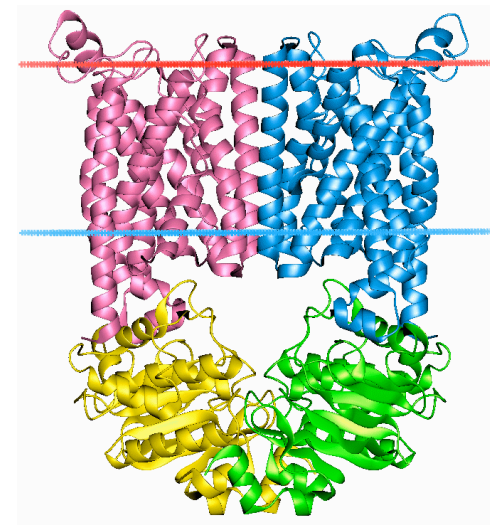
Drug Transport

Drug Transporters

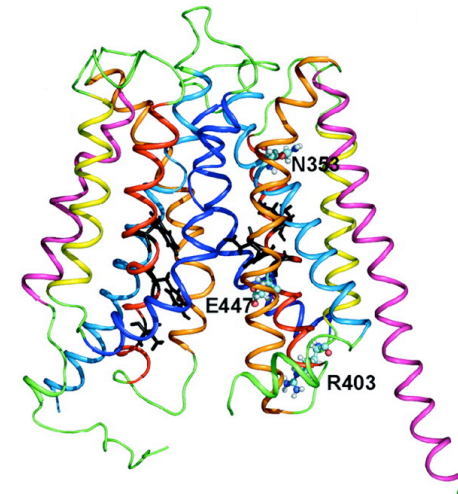
Superfamilies of Drug Transporters

- ABC Superfamily
 - ATP-Binding Cassette superfamily

- SLC Superfamily
 - Organic Anion Transporters (OATs)
 - Organic Cation Transporters (OCTs)



BtuCD



OCT2

Drug Transporters

ABC Transporters

| Transporter | Isoform | Typical Substrates | Sites in the Body |
|------------------------------------|---------|--|--|
| ABC Superfamily | | | |
| P-glycoprotein / MDR1 ^a | ABCB1 | Hydrophobic and cationic (basic) molecules; numerous drugs, including anticancer drugs | Apical surface of epithelial cell membranes. Liver, kidney, lung, intestine, BBB, testes, placenta, |
| MRP1 ^a | ABCC1 | Numerous molecules, including anticancer drugs, glucuronide and glutathione conjugates | Basolateral surface in most cell types. Kidney, lung, testes, blood:tissue barriers. |
| MRP2 ^a | ABCC2 | Numerous molecules, including anticancer drugs, glucuronide and glutathione conjugates | Apical surface of membranes. Liver, kidney, intestine |
| BRCP | ABCG2 | Anticancer, antiviral drugs, fluoroquinolones, flavinoids | Apical surface of breast ducts and lobules. Intestine, colon, liver, placenta, BBB, lungs |

From Waller D. G. et al. Medical Pharmacology and Therapeutics 3rd Edition, Saunders / Elsevier Ltd. 2010.

Drug Transporters

SLC Transporters

| Transporter | Isoform | Typical Substrates | Sites in the Body |
|------------------------|----------|---|--|
| SLC Superfamily | | | |
| OAT1 | SLC22A6 | Numerous molecules, including PAH, NSAIDs, penicillins, diuretics and phase II drug metabolites | Kidney (BL), brain, placenta, smooth muscle |
| OAT2 | SLC22A7 | Salicylate, acetylsalicylate, PGE ₂ , dicarboxylates and PAH | Kidney (BL), liver |
| OAT3 | SLC22A8 | Similar to OAT1 | Kidney (BL), liver, brain, smooth muscle |
| OAT4 | SLC22A11 | Steroid sulfate conjugates | Kidney (AP), placenta |
| OCT1 | SLC22A1 | Serotonin, noradrenaline, histamine, agmatine, aciclovir, ganciclovir | Liver, kidney, intestine, heart, skeletal muscle, placenta |
| OCT2 | SLC22A2 | Serotonin, noradrenaline, histamine, agmatine, amantidine, cimetidine | Kidney, placenta, adrenal gland, neurons, choroid plexus |
| OCT3 | SLC22A3 | Serotonin, noradrenaline, histamine, agmatine | Liver, kidney, intestine, skeletal and smooth muscle, heart, lung, spleen, neurons, placenta, choroid plexus |

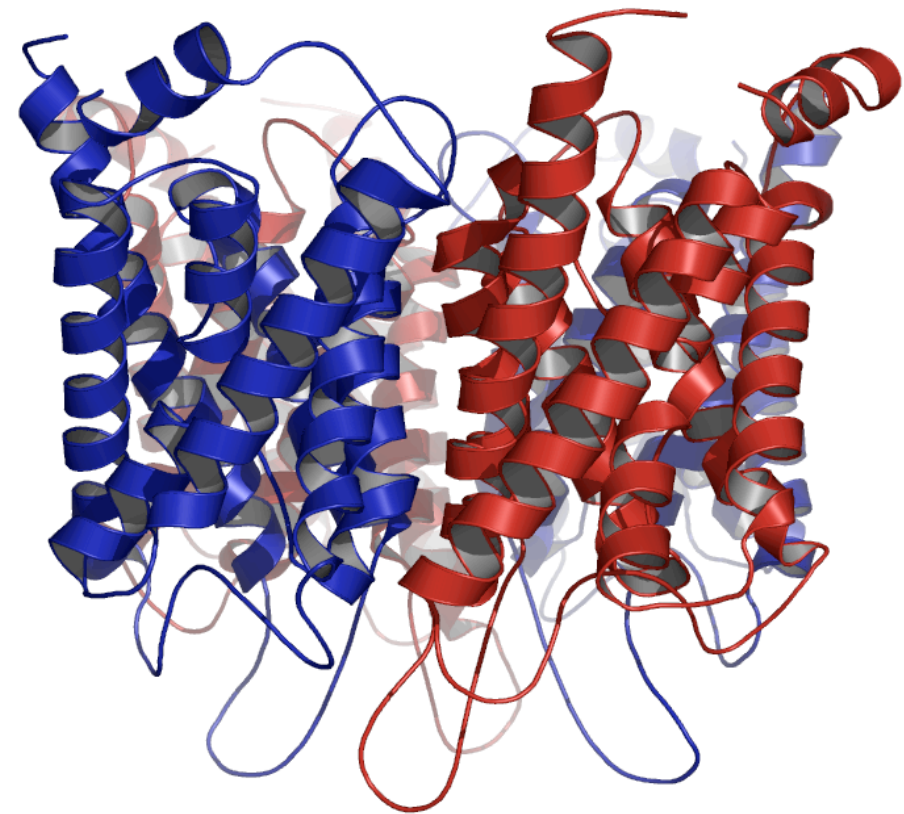
From Waller D. G. et al. Medical Pharmacology and Therapeutics 3rd Edition, Saunders / Elsevier Ltd. 2010.

Crossing Cell Membranes

Filtration

Filtration

- The movement of water and solutes down a hydrostatic pressure gradient
 - E.g, pressure exerted on cardiovascular system
- Facilitated by aquaporins and aquaglyceroporins
 - Exhibit selectivity
 - Water and small solutes only



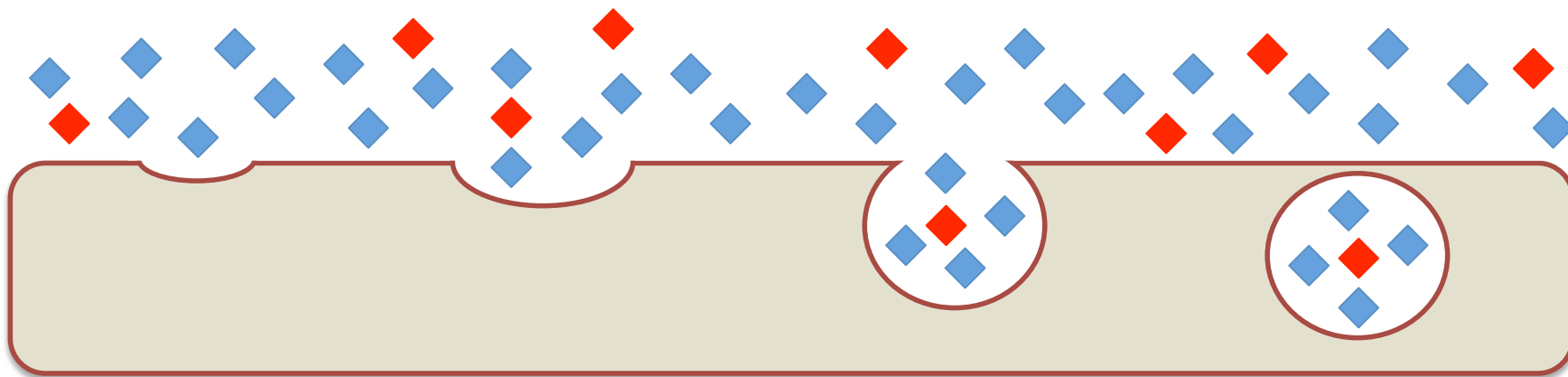
Structure of aquaporin 1 channel protein

Crossing Cell Membranes

Pinocytosis

Transport by Pinocytosis

- A form of endocytosis that allows the non-specific entry of extracellular fluid and contents to the cell by way of the formation of vesicles. It requires a usage of ATP.
- The main stages in pinocytosis are:
 - Invagination of the cell membrane
 - Pinching/fusing of the invaginated area to form a vesicle
 - Movement of the vesicle away from the cell membrane
- It is not a major route of drug transport across cellular membranes apart from protein drugs or those with appropriate formulation

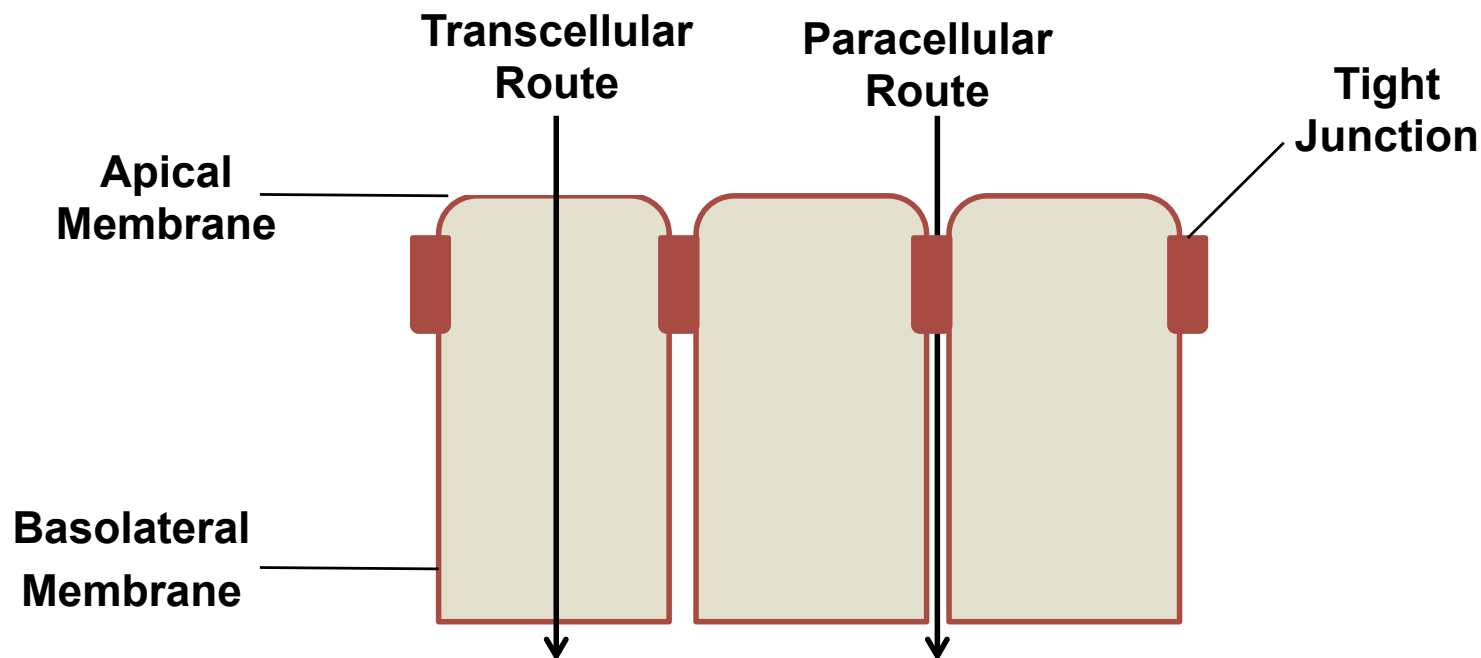


Paracellular Transport

Paracellular Transport

Transport by the Paracellular Route

- Movement of substances in between epithelial cells through the tight junctions that join them
- Most cells experience some paracellular transport
- Paracellular transport does not normally occur in the capillaries of the Blood Brain Barrier (BBB)
- Transport across the tight junctions is under cellular control / regulation



Main Points

- There are a variety of cellular membranes that are comprised largely of phospholipids and proteins
- Cell membranes consist of a phospholipid bilayer in which a variety of functional proteins can move freely (or in relation to microdomains with varying composition)
- Cell membranes serve as a barrier between adjacent aqueous compartments and control movement of electrolytes and solutes selectively or allow diffusion of molecules with appropriate physicochemical properties
- The ability of molecules to permeate the membrane by diffusion is a function of a number of factors including the relative proportions of the membrane lipid components, the physicochemical properties of the solute in question, and the transmembrane concentration gradient
- Cell membranes contain a variety of proteins that allow the passage of solutes by facilitated diffusion and active transport. Among these are ABC and SLC transporters that affect drug transport
- Other minor routes of transport such as pinocytosis and paracellular transport are also possible

